

Native Zymomonas mobilis Alcohol Dehydrogenase

Cat. No. NATE-1900

Lot. No. (See product label)

Introduction

Description Alcohol dehydrogenases (ADH) are a group of dehydrogenase enzymes that occur

in many organisms and facilitate the interconversion between alcohols and aldehydes or ketones with the reduction of nicotinamide adenine dinucleotide (NAD+ to NADH). In Humans and many other animals, they serve to break down alcohols that otherwise are toxic, and they also participate in geneRation of useful aldehyde, ketone, or alcohol groups during biosynthesis of various metabolites. In yeast, plants, and many bacteria, some alcohol dehydrogenases catalyze the opposite reaction as part of fermentation to ensure a constant supply of NAD+.

Applications The enzyme is useful for determination of alcohols or aldehydes.

Synonyms aldehyde reductase; ADH; alcohol dehydrogenase (NAD); aliphatic alcohol

dehydrogenase; ethanol dehydrogenase; NAD-dependent alcohol dehydrogenase; NAD-specific aromatic alcohol dehydrogenase; NADH-alcohol dehydrogenase; NADH-aldehyde dehydrogenase; primary alcohol dehydrogenase; yeast alcohol

dehydrogenase; EC 1.1.1.1; 9031-72-5

Product Information

Source Zymomonas mobilis

Appearance Lyophilized

EC Number EC 1.1.1.1

CAS No. 9031-72-5

Molecular Weight ca. 148,000; Subunit molecular weight: ca. 37,000

Specific Activity more than 400 U/mg protein

Contaminants (as ZM-ADH activity = 100 %) Glucose-6-phosphate dehydrogenase: < 0.10 %;

Glucokinase: < 0.02 %; Pyruvate kinase: < 0.02 %; NADH oxidase: < 0.01 %;

Lactate dehydrogenase: < 0.01 %.

pH Stability 7.0 - 9.0

Optimum pH 9.5 - 10.0

Thermal stability No detectable decrease in activity up to 40 °C.

Michaelis Constant (100 mM Glycine-KOH buffer, pH 9.0, at 30 °C) Ethanol: 110 mM; Methanol: 350

mM; NAD+: 0.12 mM; Acetaldehyde: 1.66 mM; NADH: 0.03 mM.

Specificity Ethanol: 100 %; Methanol: 0.05 %; n - Propanol: 42.3 %; n - Butanol: 0.28 %.

Unit Definition One unit of activity is defined as the amount of ZM-ADH that forms 1 μmol of NADH

per minute at 30 °C.

Reaction Alcohol + NAD+ \longleftrightarrow Aldehyde + NADH + H+

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Storage

Stable at -20 °C for at least six months.

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