Enzymatic Assay of L-ASPARTASE
(EC 4.3.1.1)

PRINCIPLE:

L-Aspartate $\xrightarrow{\text{L-Aspartase}}$ Fumarate + $\text{NH}_3$

CONDITIONS: $T = 30^\circ\text{C}$, pH 8.5, $A_{240\text{nm}}$, Light path = 1 cm

METHOD: Continuous Spectrophotometric Rate Determination

REAGENTS:

A. 150 mM Tris HCl Buffer, pH 8.5 at 30°C
(Prepare 100 ml in deionized water using Trizma Base. Adjust to pH 8.5 at 30°C with 1 M HCl.)

B. 60 mM Magnesium Sulfate Solution (MgSO$_4$)
(Prepare 1 ml in deionized water using Magnesium Sulfate, Heptahydrate)

C. 3.0 mM Ethylenediaminetetraacetic Acid Solution (EDTA)
(Prepare 1 ml in Reagent A using Ethylenediaminetetraacetic Acid, Disodium Salt, Dihydrate)

D. 500 mM L-Aspartate Solution (L-Asp)
(Prepare 10 ml in deionized water using L-Aspartic Acid, Monopotassium)

E. 5 mM Potassium Phosphate buffer, pH 7.0 at 25°C (Enz Dil)
(Prepare 25 ml in deionized water using Potassium Phosphate, Monobasic, Anhydrous
Adjust to pH 7.0 at 25°C with 1 M KOH.)

F. L-Aspartase Enzyme Solution
(Immediately before use, prepare a solution containing 2 units/ml of L-Aspartase in ice cold
Reagent E. Place on ice.)
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PROCEDURE:

Pipette (in milliliters) the following reagents into suitable cuvettes:

<table>
<thead>
<tr>
<th>Reagent</th>
<th>Test</th>
<th>Blank</th>
</tr>
</thead>
<tbody>
<tr>
<td>Reagent A (Buffer)</td>
<td>1.00</td>
<td>1.00</td>
</tr>
<tr>
<td>Reagent B (MgSO₄)</td>
<td>0.10</td>
<td>0.10</td>
</tr>
<tr>
<td>Reagent C (EDTA)</td>
<td>0.10</td>
<td>0.10</td>
</tr>
<tr>
<td>Reagent D (L-Asp)</td>
<td>0.30</td>
<td>0.30</td>
</tr>
<tr>
<td>Deionized Water</td>
<td>1.40</td>
<td>1.40</td>
</tr>
</tbody>
</table>

Mix by inversion and equilibrate to 30°C. Then add:

<table>
<thead>
<tr>
<th>Reagent</th>
<th>Test</th>
<th>Blank</th>
</tr>
</thead>
</table>
| Reagent F (Enzyme Solution) | 0.10 | -----
| Reagent E (Enz Dil)          | ---- | 0.10  |

Immediately mix by inversion and record the increase in A₂₄₀nm for approximately 5 minutes. Obtain the \( r \) A₂₄₀nm/minute using the maximum linear rate for both the Test and Blank.

CALCULATIONS:

\[
\text{Units/ml enzyme} = \frac{(r \ A_{240\text{nm}}/\text{min Test} - r \ A_{240\text{nm}}/\text{min Blank})(3)(df)}{(2.53)(0.1)}
\]

3 = Total volume (in milliliters) of assay
df = Dilution factor
2.53 = Millimolar extinction coefficient of potassium fumarate at 240 nm
0.1 = Volume (in milliliter) of enzyme used

Units/mg solid = \( \frac{\text{units/ml enzyme}}{\text{mg solid/ml enzyme}} \)

Units/mg protein = \( \frac{\text{units/ml enzyme}}{\text{mg protein/ml enzyme}} \)

UNIT DEFINITION:

One unit will convert 1.0 μmole of L-aspartate to fumarate per minute at pH 8.5 at 30°C.
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FINAL ASSAY CONCENTRATIONS:

In a 3.00 ml reaction mix, the final concentrations are 55 mM Tris, 2 mM magnesium sulfate, 0.1 mM ethylenediamine-tetraacetic acid, 50 mM L-aspartate, 0.2 mM potassium phosphate, and 0.2 unit L-aspartase.

REFERENCE:


NOTES:

1. This assay is based on the cited reference.

2. Where our Product or Stock numbers are specified, equivalent reagents may be substituted.

This procedure is for informational purposes. For a current copy of our quality control procedure contact our Technical Service Department.